



2024 JEE 27th Shift-2 Questions

39 YEARS OF ACADEMIC EXCELLENCE ASIS'S GREATEST EDUCATION BRAND IN IIT-JEE, NEET & OLYMPIADS

THE PERFECT HAT-TRICK WITH ALL-INDIA RANK





27-Jan-2024 Shift-2

Chemistry

- 1. Identify from the following species in which d^2sp^3 hybridization is shown by central atom
 - a) *SF*₆
 - b) BrF_5
 - c) $[Pt(Cl_4)]^{-2}$
 - d) $[CO(NH_3)_6]^{+3}$

Ans: d

- 2. Which type of protein can not be denaturized when heated
 - a) Primary
 - b) Secondary
 - c) Tertiary
 - d) Quaternary
 - Ans: d
- 3. Phenolic group can be identified by a positive:
 - (a) Phthalein dye test, (b) Carbylamine test, (c)Lucas test (d) Tollen's test. Ans: a
- 4. Incorrect pair form
 - a) Habber process Iron
 - b) Polythene $\frac{\text{Ticl}_4}{\text{Al}(\text{CH}_3)_3}$

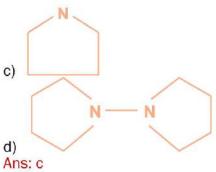
 - c) Photography AgBr
 - d) Wacker process PtCl₂

Ans: d

5.
$$Cl - (CH_2)_4 - Cl \xrightarrow{excess NH_3} \xrightarrow{NaOH} A + H_2O + Nacl$$

a) $Cl^- + NH_3 - (CH_2)_4 - NH^+Cl^-$
b) $NH_2 - (CH_2)_4 - NH_2$





- 6. The quantity which changes with temperature:
 - a) Mole fraction
 - b) Mass Percentage
 - c) Molarity
 - d) Molality
 - Ans: c
- 7. Which of the following can not act as an oxidizing agent?
 - (a) MnO_4^-
 - (b) SO_4^{2-}
 - (c) N^{3-}
 - (d) BrO_3^-
 - Ans: c
- 8. Phenolic group can be identified by a positive
 - a) Lucas test
 - b) Carbylamine test
 - c) Phthalein test
 - d) Tollen's test

Ans: c

9. Products for the below reaction are:



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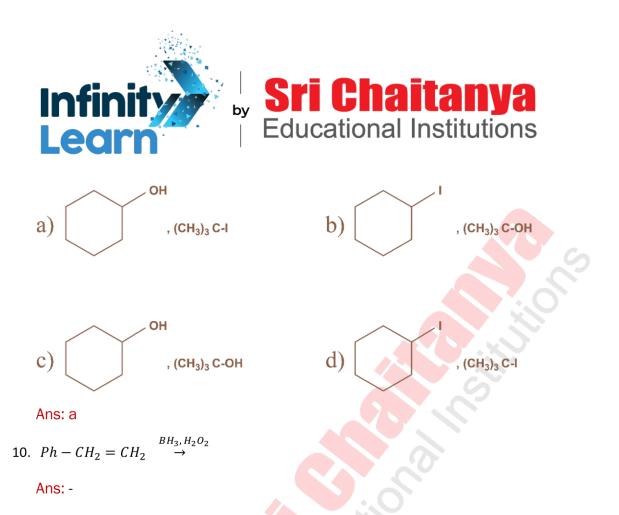


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11. No. of non-polar compounds H₂O, CO₂, NH₃, BF₃, CH₄, SO₂, HF, HCl Ans: 3

12. No. of compounds with chiral centres NO_2 OH H CH - CH₃, CH₃ - CH - CH₂ - OH, CH₃ - CH - CH₂ - CH₂I Ans: 4

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- 13. Which of the options, all the elements have d^{10} configuration in their ground state
 - a) Cu, Zn, Cd, Ag
 - b) Cd, Au, Hg, Ni
 - c) Sc, Ti, Fe, Zn
 - d) Fe, Cr, Co, Ni Ans: a

14. Steam volatile & water immiscible compounds can be separated by

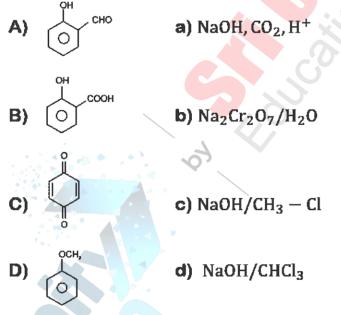
- a) Distillation
- b) Fractional distillation

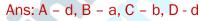


- c) Distillation under reduced pressure
- d) Steam distillation Ans: d
- 15. For a first order reaction $t_{99.9\%} = x t_{50\%}$. Find out the value of x Ans: 10
- 16. Which of the following will not give $S_{\rm N} {\rm 1}$
 - a) $CH_2 = CH CH_2 Cl$
 - b) $Ph CH_2 Cl$
 - c) $CH_3 CH = CH Cl$
 - d) $(H_3C)_3C Cl$

Ans: c

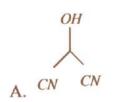
17. Match the following

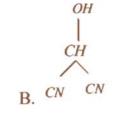


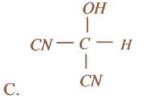


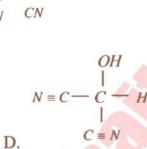
18. Which represents the line formula of (OH)CH (CN)₂











Ans: a



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Maths

- 1. 20th term from the end of $20,19\frac{1}{4}, 18\frac{1}{2}, 17\frac{3}{4}, \dots, -129\frac{1}{4}$ Ans: (-115)
- 2. The position vector of vertices A, B, C of Δ are î + 2ĵ + 3k̂, î + ĵ + 3k̂, 2î + ĵ + 3k̂
 respectively. Let l is the length of angle bisector of ∠BAC, then the value of l² is:
 (a) 4 + 2√2
 - (b) $4 2\sqrt{2}$ (c) $2 + 2\sqrt{2}$ (d) $2 - 2\sqrt{2}$ Ans: (b)
- 3. If *A* is a 2 × 2 matrix and *I* is an Identity matrix of order 2 $\&|A - \lambda I| = 0$ gives values of λ as -1&3. Then, trace of A^2 is equal to Ans: (10)
- 4. If $\frac{dy}{dx} = \frac{x+y-2}{x-y}$, and y(0) = 2, find y(2). (a) 0 (b) 2 (c) e (d) e^2
 - Ans: (a)
- 5. The area bounded by $0 \le y \le \min\{2x, 6x x^2\}$ and x-axis is A. then 12 A is: Ans: (304 sq. Units)

6. 0 < a < 1, $\int_{0}^{\pi} \frac{dx}{1-2a\cos x+a^{2}} =$ value of the integral (a) $\frac{\pi^{2}}{\pi + a^{2}}$ (b) $\frac{\pi}{1 + a^{2}}$

(b) $\frac{\pi}{1+a^2}$ (c) $\frac{\pi^2}{\pi-a^2}$ (d) $\frac{\pi}{1-a^2}$







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- 7. The integral $\int \frac{(x^8 x^2)}{(x^{12} + 3x^6 + 1)\tan^{-1}(x^3 + \frac{1}{x^3})} dx$ is equal to: (a) $\frac{1}{3} \ln \left| \left(\tan^{-1} \left(x^3 + \frac{1}{x^3} \right) \right) \right| + C$ (b) $\ln \left| \left(\tan^{-1} \left(x^3 + \frac{1}{x^3} \right) \right) \right| + C$ (C) $\frac{1}{6} \ln \left| \left(\tan^{-1} \left(x^3 + \frac{1}{x^3} \right) \right) \right| + C$ (d) $\frac{1}{9} \ln \left| \left(\tan^{-1} \left(x^3 + \frac{1}{x^3} \right) \right) \right| + C$ Ans: (a)
- 8. If $2\tan^2 \theta 5\sec \theta = 1$ has exactly 7 solutions in $\left[0, \frac{n\pi}{2}\right]$ for least value of $n \in N$, then $\sum_{k=1}^{n} \frac{k}{2^n}$ is equal to
 - (a) $\frac{9}{2^9}$ (b) $\frac{91}{2^{13}}$ (c) $\frac{7}{2^7}$ (d) $\frac{11}{2^{12}}$ Ans: (b)
- 9. Let *R* be the interior region between the lines 3x y + 1 = 0 and x + 2y 5 = 0 containing the origin. The set of all values of *a* for which the points $(a^2, a + 1)$ lies in *R* is
 - (a) $(-\infty, -1) \cup (3, \infty)$
 - (b) $(-3,0) \cup \left(\frac{1}{3},1\right)$
 - (C) $(-\infty, -1) \cup \left(0, \frac{1}{3}\right)$
 - (d) $(-\infty, -2) \cup \left(0, \frac{1}{3}\right)$

Ans: (b)

10. If $f(x) = 6x - x^2$, $x \in [0,2]$ and $g(x) = \begin{cases} \min f(t), 0 \le t \le x \\ 3 + x, x \in [1,2] \end{cases}$ then number of points where g(x) is not differentiable is: (a) 1 (b) 0 (c) 2 (d) 3 Ans: (a)





- 11. If for two sets A&B, n(A) = m and n(B) = n. Also, (Number of subsets of A Number of subsets of B) = 56. Then find the value of (2m + n) Ans: (15)
- 12. If the line x + y = 0 is tangent to the circle $(x \lambda)^2 + (y \beta)^2 = 50$, then $(\lambda + \beta)^2 =$ Ans: (100)
- 13. If $f(x) = \int_0^x g(t) \ln\left(\frac{1-t}{1+t}\right) dt$ and g is odd continuous function and $\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} \left(f(x) + \frac{x^2 \cos x}{(1+e^x)}\right) dx = \frac{\pi^2}{\alpha^2} \alpha$, then α is Ans: (-2)
- 14. If α and β are the roots of the equation $x^2 x 1 = 0$ and $S_n = 2024 \cdot \alpha^n + 2024 \cdot \beta^n$. Then S_3 is equal to
 - (a) 8096
 - (b) 4048
 - (C) 1024
 - (d) 2024
 - Ans: (a)
- 15. If the mean of 15 observations is 12 and standard deviation is 3. If 12 is replaced by 10 in data, then the new mean is μ and variance is σ^2 , then what is the value of $15(\mu + \mu^2 + \sigma^2) =$ Ans: (2429)
- 16. If $\lim_{x\to 0} \frac{3+\alpha \sin x + \beta \cos x + \ln(1-x)}{3\tan^2 x} = \frac{1}{3}$, then $2\alpha \beta$ is equal to Ans: (5)





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27-Jan-2024 Shift-2

Physics

- 1. The threshold frequency of metel with work function 6.63ev is Ans: $1.6 \times 10^{15} Hz$
- 2. Does kinetic friction and static friction depend on area of contact and material of surface.
 - a) only on surface area
 - b) only on material
 - c) both material on surface
 - d) None of these
 - Ans: b
- Total kinetic energy of 1 mol of oxygen at 27°C Ans: 6250 J
- 4. Current of 200 A deflects the coil of a moving coil galvanometer by 60°. Find the current to cause deflection of $\frac{\pi}{10}$ rad

Ans: 60 A

5. During an adiabatiic process the pressure of gas is proportional to cube of its absolute temperature the ratio of C_p/C_v is

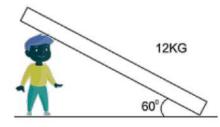
Ans: $\frac{3}{2} = 1.5$

- A bullet is fired into a wooden plank , its velocity becomes 1/3 when it penetrates by 4 cm how much further it should penetrate for its velocity to become zero Ans: 0.5 cm
- 7. A body is allowed to free fall from height h .Distance between two points A & B in the motion is 80m and time taken to travel from A to B is 2sec . Find the distance of point A from the initial point of projection Ans: 45 m
- 8. A body is projected up with a velocity to $3^{th}/4$ of the escape velocity from The surface of the earth. The height it reaches from the centre of the earth is



Ans: $\frac{9R}{7}$

9. Find the normal contact force between man and rod



Ans: 30 N

- 10. In a Wheatstone bridge of resistance X with specific resistance = $\frac{x\pi r^2}{L}$ = Sr when the length is doubled What is the new specific resistance
 - (a) $S_1/2$ (b) 2 S₁ (c) 4 S₁ (d) S_1
 - Ans: d
- Assertion: for equipotential surface work done by the field is 0 Reason: electric lines of force is perpendicular to the surface Ans: A – T & R - T
- 12. Longest wavelength for paschen series is given by $\lambda = \frac{\alpha}{7R_H}$. Then value of α is Ans: $\alpha = 144$



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