



2024 JEE 29th Shift-1 Questions

39 YEARS OF ACADEMIC EXCELLENCE ASIS'S GREATEST EDUCATION BRAND IN IIT-JEE, NEET & OLYMPIADS

THE PERFECT HAT-TRICK WITH ALL-INDIA RANK





PHYSICS

SECTION - A

- Debroglie wavelength of an electron & photon is same. Velocity of electron is 25% of velocity of light. Find the ratio of Kinetic Energy of electron by photon?
 - (1) 1
 - (2) 1/2
 - (3) 8
 - *(4)1/8
- 2. The voltage applied across the resistance R is V= (200 \pm 5) Volts and current in the resistance is i = (20 \pm 0.5) Amp. Find % error in Resistance?
 - (1) 3.5%
 - *(2)5%
 - (3) 7%
 - (4) 3%
- 3. Find the ratio of total Kinetic Energy in SHM when x = A/3, where (A) is amplitude.
 - $(1) \ 3/4$
 - (2) 4/3
 - (3) 8/9
 - *(4)9/8
- 4. A solid cylinder is released from rest. Surface is rough enough for rolling on inclined plane of angle of inclination of $\theta = 60^{\circ}$. Find (a_{com}) of the cyclinder?
 - (1) 10/3
 - *(2) 10/ $\sqrt{3}$
 - (3) 10 \sqrt{3}
 - (4) 5
- 5. A capacitor of capacitance is 100μ F is charged to a potential of 12V and 6.4 mH inductor is connected to it. Find maximum value of current.
 - *(1) 1.5 A
 - (2) 20. A
 - (3) 1.2 A
 - (4) 3.2 A
- 6. A galvanometer having coil resistance 10 $_\Omega\,$ shows a total scale deflection for current of 3mA. For it to

measure a current of 8A the value of shunt should be

(1) $2.75 \times 10^{-3} \Omega$ *(2) $3.75 \times 10^{-3} \Omega$ (3) $3 \times 10^{-3} \Omega$ (4) $4.85 \times 10^{3} \Omega$

- 7. Given below are two statements.
 - Statement I : If a capillary tube is immersed first in cold water and then in hor water the height of capillary rise will be smaller in hot water
 - Statement II : If a capillary tube is immersed first in cold water and then in hot water capillary reise will be smaller in cold water.
 - (1) Both statement I and statement II are correct.
 - (2) Both statement I and statement II are incorrect.
 - *(3) Statement I is correct but statement II is incorrect.
 - (4) Statement I is incorrect but statement II is correct.
- 8. If A biconvex lens of refractive index 1.5 has a focal length of 20 cm in air. Its focal length when immersed in a liquid of refractive index 1.6 will be
 - (1) +16 cm
 - (2) +160 cm
 - *(3)-160 cm
 - (4) -16cm
- 9. If the ratio of centripetal acceleration of two particles moving on the same circular path is 3: 4. Find the ratio of their speed.
 - (1) 2: $\sqrt{3}$
 - *(2) √3:2
 - (3) $\sqrt{3}:1$
 - (4) $\sqrt{2}:1$
- 10. If an object is having same weight at some distance d above and below the surface of earth. Find its distance from surface of earth.

(1)
$$\frac{R}{2}$$

*(2) $\frac{(\sqrt{5}-1)R}{2}$

1

$$(3) \quad \frac{(\sqrt{3}-1)R}{2}$$

- (4) $(\sqrt{5}-1)R$
- 11. A stationary hydrogen atom de excites from first excited state to ground state. The recoil speed of hydrogen atom is n ×10⁸ m/s. Find the value of n up to nearest integer value.(mass of hydrogen atom = 1.8×10^{-27} kg)
 - = 2
- 12. A body of mass 100 kg travelled 10m before coming to rest. If μ = 0.4, work done against friction is -(motion is happening on a horizontal surface, take g = 10m/ s^2)
 - (1) 4500 J
 - (2) 50000 J
 - (3) 4200 J
 - *(4) 4000 J
- A solid sphere of radius 4a units is placed with its centre at origin. Two charges -2q at (-5a, 0) & 5q at (3a, 0) is

placed. If the flux through the sphere is $\frac{xq}{\varepsilon}$, find x?

= 5

CHEMISTRY

- 1. Which of the following pair will be formed by the decomposition of $KMnO_4$.
 - (1) MnO₄, MnO₂
 - $(2) K_2 MnO_4, MnO_2$
 - (3) KMnO₄, MnO₂
 - (4) MnO₂, H₂O

Answer: (b) K₂ MnO₄ , MnO₂

Solution:

Potassium permanganate forms dark purple (almost black) crystals which are isostructural with those of KCIO_4 . The salt is not very soluble in water (6.4 g / 100 g of water at 293 K), but when heated it decomposes at 513 K

 $2KMnO_4 \rightarrow K_2MnO_4 + MnO_2 + O_2$

- 2. Interaction between π Bond & lone pairs on adjacent atoms
 - (a) Resonance
 - (b) Hyper conjugation

- (c) Inductive Effect
- (d) Electronic Effect

Answer: (a) Resonance

3. Assertion (A) : Electronegativity increase across a period

Reason (**R**) : Effective increase in nuclear charge is more than effective shielding.

Solution: Assertion true reason true

4. S-1: Electronegativity increase down the group 14 is to pb

S-2: Group 14 contains metals, non metals and also metalloids

S-1 is incorrect but S-2 is correct

Colu	ımn -	I N		Column -	II
a) Zi	egler N	atta Cat	alyst	i) Rh	
b) Bl	ood Pig	gment		ii) CO	
c) Wi	ilkinsor	n Catalys	st	iii) Fe	
d) Vi	tamin E	B ₁₂		iv) Ti	
	(a)	(b)	(c)	(d)	
(1)	iv	iii	i	ii	
(2)	iii	ii	i	iv	
(3)	ii	iv	i	ii	
(4)	i	ii	iii	iv	
Key:	1				

- Appearance of Red colour on treatment with Na fusion extract of an organic compound with FeSO₄ in presence of conc. H₂SO₄ indicate element
 - a) N
 - b) Br
 - c) S
 - d) N&S

Answer: (d) N & S

6. Cl⁻ shows disproportionation in alkaline medium:

acl_2	+ b OH	\rightarrow c cl	0⁻ + dcl	- + H ² O
	(a)	(b)	(c)	(d)
(1)	1	1	1	3
(2)	3	6	2	4
3)	1	2	1	1



1.H2O/H

2.CrO₃/H⁺ 3.NH₂-NH₂,KOH,Δ

A

JEE Main 29_01_(shift 01)- Physics 2 Benzaldehyde, acetophenone, methanal 4) 4 1 3 Answer: (b) Key:1 Solution: $3O_2 + 60H^- \rightarrow 2CIO_3^- + 4CI^- + 3H_2O^-$ 7. The correct set of four Quantum numbers of Valence 15. electron of Rb(37) a) n = 5 ., l = 0 ., m = 1., b) n = 5., l = 0., m = 0., Key: c) n = 5., l = 1., m = 0., d) n = 5., l = 1., m = 1., Answer: b Type of amino acids obtained on hydrolysis of proteins b) β a) α 17. c) γ d) δ Which of the follwing coordinatin compounds has bridging carbonyl ligand a) [Mn₂ (CO)₁₀ b) [Co₂ (CO)₈] a c) $[Cr(CO)_{6}]$ d) [Fe(CO)₅] 10. Calculate the Molarity of a Solution having density = 1.25 g/ml. % (w/w) of Solute is 31.4% of H₂SO₄ solution a) 4 b) 9 c) 8 d) 6 11. Among the hetercyclic compound that contain sulphur atom is a) Pyradizine b) Furan c) Thophene d) Pyrrole Key:c d 12. Find the weight of Zinc formed when Zinc sulphate is electrolized for 15 min by using 0.015 Ampere current Key: 0.0046g 18. 13. Number of compound in which B.0 = 1 and is paramagnetic $He_2^+O_2^+, O_2^{-2}, N_2^+$

8.

9.

alkline medium 16. $MnO_{4}^{-} + KI$ >?+10⁻₃ Key: MnO₂ Cl₂ $\xrightarrow{CCl_4} B$ CH₃ Conc.HBr

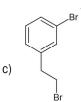
14. Number of compounds that gives positive fehling test

3

Key:0







- 19. Number of compunds with central atom having lone pair of electrons.
 - 0₃, NH₃, SF₄, H₂O, XeF₄
- 20. The reactions taking place in temperature, the rate of overall reaction is given as $K = \frac{k_1 K_2}{k_3}$ then overall activate Ea for reaction, Ea₁ = 40 kg / mole, Ea₂ = 50 kg mole, Ea₃ = 60 kg / mole
- 21. Which of the follwing is correctly matches
 - a) Cryolite Na₃AIF₆
 - b) Flurospar CaF₂
 - c) Chloropatite $-3Ca_3 (PO_4)_5 CaF_2$
 - d) Carnalite KCI.MgCl₂ . 6H₂0
 - Key:3

MATHEMATICS

1. If
$$\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} \left(\frac{x^2 \cos x}{1 + e^x} + \frac{1 + \sin^2 x}{1 + e^{\sin x^{2023}}} \right) dx = \frac{\pi}{2} (\pi + \alpha) - 4 \text{ find}$$

 α .

Ans : -12

2. A position vector of A is $\overrightarrow{OA} = 2i + 2j + k$, $\overrightarrow{OB} = 2i + 4j + 4k$ Then angular bisector of $\angle AOB$ intersect AB at point C. Find \overrightarrow{OC} .

Ans:
$$\begin{pmatrix} 2, \frac{8}{3}, 2 \end{pmatrix}$$

3. If $\begin{bmatrix} 1 & 0 & 0 \\ 0 & \alpha & \beta \\ 0 & \beta & \alpha \end{bmatrix}$ and $|2A|^3 = 2^{21}$. Where $\alpha, \beta \in N$

Then (α, β) can be

Ans : (3,5)

4. Sum of all 64 terms of G.P is 7 times of odd terms then find common ratio.

Ans: $\frac{1}{7}$

5. Let a die rolled till 2 is obtained. The probability that 2 obtained on even numbered toss is equal to :

$$ns: \frac{5}{11}$$

Δ

6. Find the common ratio of a G.P whose $a_6 = 2$ and

 $a_1 a_4 a_7$ is maximum.

Ans : $\frac{8}{5}$

7. If a, β are the roots of $x^2-x+2=0$ such that $Im(\alpha) > Im(\beta)$, find $\alpha^6 + \alpha^4 + \beta^4 - 5\alpha^2$

Ans : 13

8. Given $\Delta^{le}ABC$, y=x is angle bisector of $\angle ABC 2x-y=2$ is equation if AC(4,6) and (α, β) are A, B respectively, then $\alpha + 2\beta = ?$

9. Evaluate : $\lim_{x \to \frac{\pi}{2}} \frac{\int_{x}^{(\frac{\pi}{2})^{3}} \cos t^{\frac{1}{3}} dt}{\left(x - \frac{\pi}{2}\right)^{2}}$ (1) $\frac{3\pi^{2}}{4}$

(2)
$$\frac{3\pi}{4}$$





(3)
$$\frac{3\pi^2}{8}$$

(4)
$$\frac{3\pi}{8}$$

Ans :
$$\frac{3\pi^2}{8}$$

10.
$$\frac{11}{2}c_1 + \frac{11}{3}c_2 + \dots + \frac{11}{10}c_9 = \frac{m}{n}$$
 The m+n is

Ans : 2041

11. Rank of the word 'GTWENTY' in dictionary is

Ans : 553

12. $\overline{a}, \overline{b}, \overline{c}$ are three non collinear vectors $\overline{a}, 6\overline{b}$, is collinear with $\overline{c}, \overline{b} + 5\overline{c}$ is collinear with \overline{a} . then $\overline{a} + \alpha \overline{b} + \beta \overline{c} = 0$ then find $\alpha + \beta = ?$

Ans : 36

13. If
$$f(x) = \int \frac{\cos ec x + \sin x}{\cos ec x \sec x + \tan x \sin^2 x} dx$$

and
$$\lim_{x \to \frac{\pi}{2}} f(x) = 1$$
, then find $f\left(\frac{\pi}{4}\right) = ?$



- $\frac{1}{2}A\left[\left(A+A'\right)\right]^{2}+\left(A-A'\right)^{2}\right]$ equal to
- (1) A³+A^T
- (2) A²+A^T
- (3) A²+I
- (4) A³+I.
- Ans : A³+A^T

15. If
$$z = \frac{1}{2} - 2i$$
 such that
 $|z+1| = \alpha z + \beta (1+i), i = \sqrt{-1}$ and $\alpha, \beta \in N$ then
 $\alpha + \beta$ is
Ans:

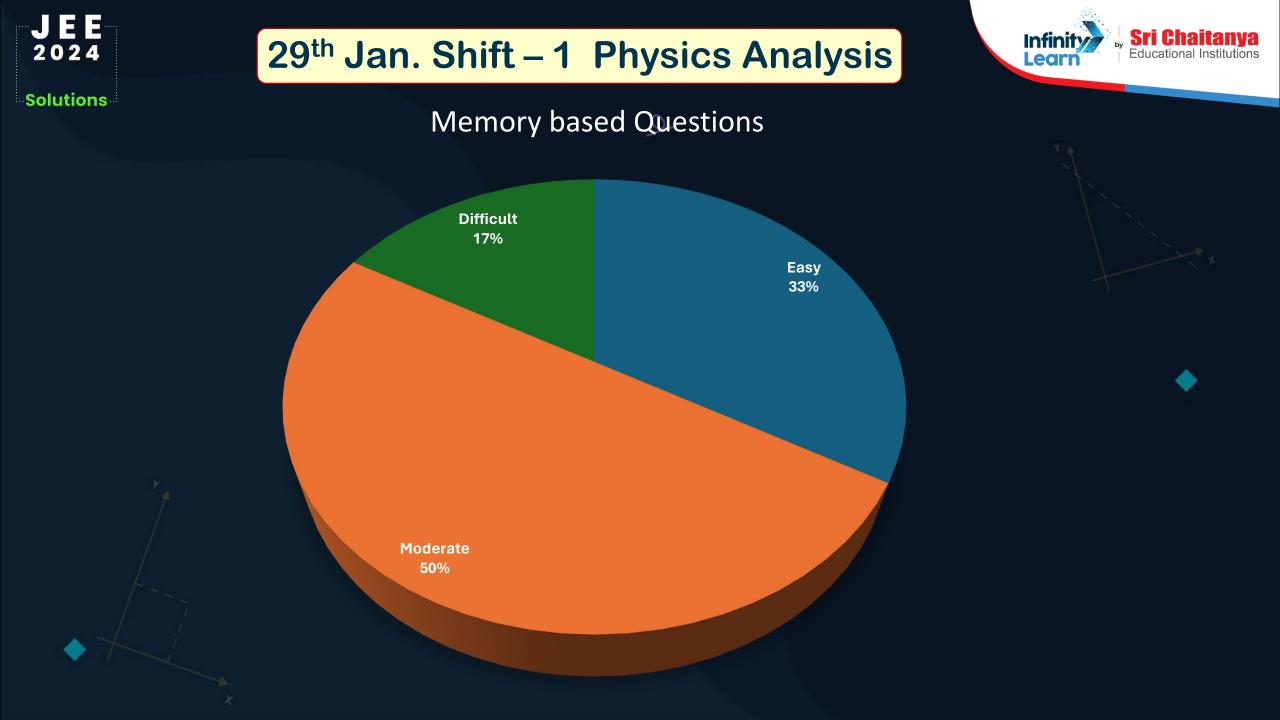
16. Let R be a relation on $z \times z$ defined by (a,b)(c,d) if adbc is divisible by 5 is what type of relation?

17. Domain of
$$f(x) = \frac{\log(x-1)}{\log_{x-1}(x-4)}$$
 is

18. If
$$\frac{dy}{dx} - \left(\frac{\sin 2x}{1 + \cos^2 x}\right)y = \frac{\sin x}{1 + \cos^2 x}$$
 and $y(0)=0$ then $y\left(\frac{\pi}{2}\right)$ is

Ans:1

Ans: (4,∞)



29th Jan. Shift – 1 Physics Analysis



Chapter	Grade	No. of Questions	Difficulty Level
Units & Measurement	11	1	Easy
Motion in a straight line	11	1	Moderate
Motion in a plane	11	1	Easy
Laws of motion	11	2	Moderate, Easy
Work, energy and power	11	2	Moderate, Easy
System of particles and rotational motion	11	1	Moderate
Gravitation	11	1	Moderate
Mechanical properties of solids	11		
Mechanical properties of fluids	11	1	Moderate
Thermal properties of matter	11 🔷		

29th Jan. Shift – 1 Physics Analysis



Chapter	Grade	No. of Questions	Difficulty Level
Thermodynamics	11	1	Easy
Kinetic theory of gases	11	1	Easy
Oscillations	11	1	Easy
Waves	11		
Electrostatic potential and capacitance	12	1	Moderate
Electric charges and fields	12	1	Difficult
Current Electricity	12	2	Easy, Difficult
Moving charges and magnetism	12	1	Moderate
Magnetism and matter	12		
Electromagnetic induction	12	1	Moderate

29th Jan. Shift – 1 Physics Analysis

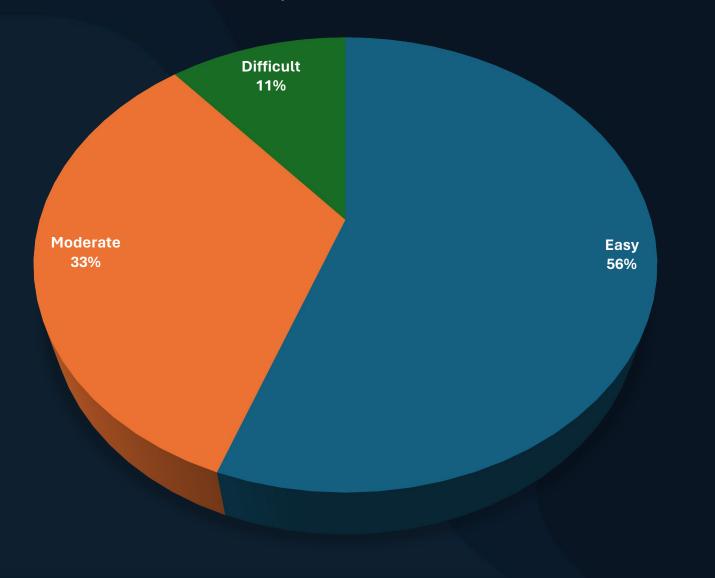
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Chapter	Grade	No. of Questions	Difficulty Level
Alternating Current	12	1	Difficult
Electromagnetic waves	12		
Ray optics and optical instruments	12	2	Moderate
Wave optics	12		
Dual nature of radiation and matter	12	1	Moderate
Atoms	12	1	Moderate
Nuclei	12		
Semiconductor electronics: Materials; devices and simple circuits	12		



29th Jan. Shift – 1 Maths Analysis





29th Jan. Shift – 1 Maths Analysis



Chapter	Grade	No. of Questions	Difficulty Level
Sets and Relations	11	1	Easy
Functions	11	1	Easy
Trigonometric Functions	11		
Principle of Mathematical Induction, Linear inequalities	11		
Quadratic Equations	11		
Complex Numbers	11	2	Easy
Permutations and Combinations	11	1	Easy
Binomial Theorem	11	1	easy
Sequences and Series	11	1	Moderate
Straight Lines	11	1	Moderate

29th Jan. Shift – 1 Maths Analysis

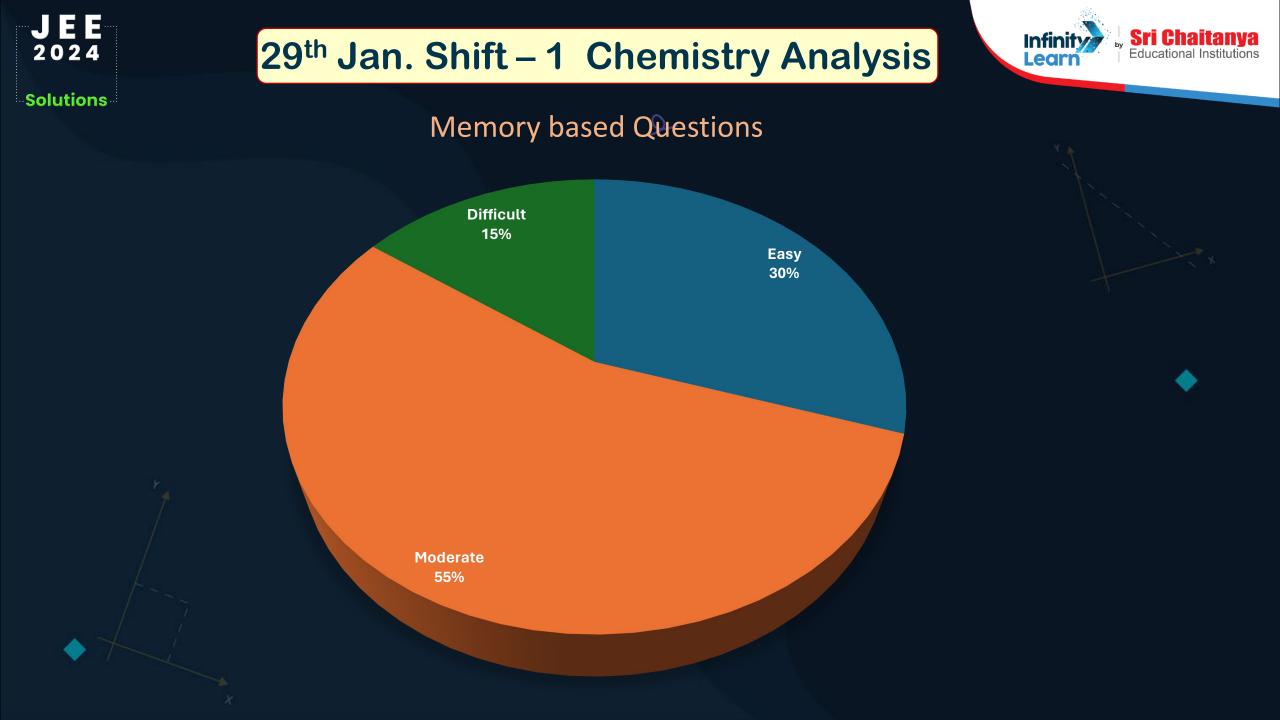


Chapter	Grade	No. of Questions	Difficulty Level
Circles	11		
Conic Section	11		
Limits and Derivatives	11		
Statistics	12		
Matrices	12	2	Easy
Determinants	12		
Inverse Trigonometric Functions	12		
Continuity and Differentiability	12		
Applications of Derivatives	12	1	Difficult
Indefinite Integration	12	1	Moderate

29th Jan. Shift – 1 Maths Analysis



Chapter	Grade	No. of Questions	Difficulty Level
Definite Integration	12	2	Moderate
Applications of Integrals	12		
Differential Equations	12		
Vectors Algebra	12	2	Easy
Three Dimensional Geometry	12		
Probability	12	1	Moderate
Trigonometric Equations	11		



29th Jan. Shift – 1 Chemistry Analysis



Chapter	No. of Questions	Difficulty Level
Some basic concepts of chemistry	1	Moderate
Structure of Atom	1	Moderate
Periodic Classification		
Chemical bonding and molecular structure	3 ×	Moderate, Easy
States of matter		
Thermodynamics		
Equilibrium		
Redox reactions	3	Moderate, Easy
p - block elements	2	Easy
Hydrocarbons	2	Difficult, Moderate

29th Jan. Shift – 1 Chemistry Analysis



Chapter	No. of Questions	Difficulty Level
Solution		
Electrochemistry	1	Moderate
Chemical kinetics	1	Moderate
General principles and processes of isolation of elements	-t×	
d & f block elements	1	Moderate
Coordination compounds	1	Moderate
Haloalkanes and Haloarenes	1	Difficult
Alcohol, Phenol Ether	2	Moderate
Aldehyde, Ketone, Carboxylic acid		
Amines		

29th Jan. Shift – 1 Chemistry Analysis



Chapter	No. of Questions	Difficulty Level
Biomolecule	1	Easy
Organic Chemistry- Some basic principles and techniques(General Organic)	2	Easy
Salt Analysis		







HISTORY CREATED

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