# 2024 JEE $29^{\text {th }}$ Shift-1 Questions HISTORY CREATED <br> <br> 39 YEARS OF ACADEMIC EXCELLENCE <br> <br> 39 YEARS OF ACADEMIC EXCELLENCE ASIS'S GREATEST EDUCATION BRAND IN IIT-JEE, NEET \& OLYMPIADS 

## THE PERFECT HAT-TRICK WITH ALL-INDIA RANK 1



## PHYSICS

## SECTION - A

1. Debroglie wavelength of an electron \& photon is same. Velocity of electron is $25 \%$ of velocity of light. Find the ratio of Kinetic Energy of electron by photon?
(1) 1
(2) $1 / 2$
(3) 8
*(4) 1/8
2. The voltage applied across the resistance $R$ is $V=(200$ $\pm 5)$ Volts and current in the resistance is $i=(20 \pm 0.5)$ Amp. Find \% error in Resistance?
(1) $3.5 \%$
*(2) $5 \%$
(3) $7 \%$
(4) 3\%
3. Find the ratio of total Kinetic Energy in SHM when $x=$ $A / 3$, where (A) is amplitude.
(1) $3 / 4$
(2) $4 / 3$
(3) $8 / 9$
*(4) $9 / 8$
4. A solid cylinder is released from rest. Surface is rough enough for rolling on inclined plane of angle of inclination of $\theta=60^{\circ}$. Find ( $\mathrm{a}_{\text {com }}$ ) of the cyclinder?
(1) $10 / 3$
*(2) 10/ $\sqrt{3}$
(3) $10 \sqrt{3}$
(4) 5
5. A capacitor of capacitance is $100 \mu \mathrm{~F}$ is charged to a potential of 12 V and 6.4 mH inductor is connected to it. Find maximum value of current.
*(1) 1.5 A
(2) 20. A
(3) 1.2 A
(4) 3.2 A
6. A galvanometer having coil resistance $10 \Omega$ shows a total scale deflection for current of 3 mA . For it to
measure a current of 8 A the value of shunt should be
(1) $2.75 \times 10^{-3} \Omega$
*(2) $3.75 \times 10^{-3} \Omega$
(3) $3 \times 10^{-3} \Omega$
(4) $4.85 \times 10^{3} \Omega$
7. Given below are two statements.

Statement I: If a capillary tube is immersed first in cold water and then in hor water the height of capillary rise will be smaller in hot water

Statement II: If a capillary tube is immersed first in cold water and then in hot water capillary reise will be smaller in cold water.
(1) Both statement I and statement II are correct.
(2) Both statement I and statement II are incorrect.
*(3) Statement I is correct but statement II is incorrect.
(4) Statement I is incorrect but statement II is correct.
8. If A biconvex lens of refractive index 1.5 has a focal length of 20 cm in air. Its focal length when immersed in a liquid of refractive index 1.6 will be
(1) +16 cm
(2) +160 cm
*(3) -160 cm
(4) -16 cm
9. If the ratio of centripetal acceleration of two particles moving on the same circular path is 3 : 4 . Find the ratio of their speed.
(1) $2: \sqrt{3}$
*(2) $\sqrt{3}: 2$
(3) $\sqrt{3}: 1$
(4) $\sqrt{2}: 1$
10. If an object is having same weight at some distance $d$ above and below the surface of earth. Find its distance from surface of earth.
(1) $\frac{R}{2}$
*(2) $\frac{(\sqrt{5}-1) R}{2}$
(3) $\frac{(\sqrt{3}-1) R}{2}$
(4) $(\sqrt{5}-1) R$
11. A stationary hydrogen atom de excites from first excited state to ground state. The recoil speed of hydrogen atom is $n \times 10^{8} \mathrm{~m} / \mathrm{s}$. Find the value of $n$ up to nearest integer value. (mass of hydrogen atom $=1.8 \times$ $10^{-27} \mathrm{~kg}$ )
$=2$
12. A body of mass 100 kg travelled 10 m before coming to rest. If $\mu=0.4$, work done against friction is -(motion is happening on a horizontal surface, take $g=10 \mathrm{~m} /$ $\mathrm{s}^{2}$ )
(1) 4500 J
(2) 50000 J
(3) 4200 J
*(4) 4000 J
13. A solid sphere of radius $4 a$ units is placed with its centre at origin. Two charges $-2 q$ at $(-5 a, 0) \& 5 q$ at $(3 a, 0)$ is placed. If the flux through the sphere is $\frac{x q}{\varepsilon_{0}}$, find $x$ ? $=5$

## CHEMISTRY

1. Which of the following pair will be formed by the decomposition of $\mathrm{KMnO}_{4}$.
(1) $\mathrm{MnO}_{4}, \mathrm{MnO}_{2}$
(2) $\mathrm{K}_{2} \mathrm{MnO}_{4}, \mathrm{MnO}_{2}$
(3) $\mathrm{KMnO}_{4}, \mathrm{MnO}_{2}$
(4) $\mathrm{MnO}_{2}, \mathrm{H}_{2} \mathrm{O}$

Answer: (b) $\mathrm{K}_{2} \mathrm{MnO}_{4}, \mathrm{MnO}_{2}$
Solution:
Potassium permanganate forms dark purple (almost black) crystals which are isostructural with those of $\mathrm{KClO}_{4}$. The salt is not very soluble in water (6.4 $\mathrm{g} / 100 \mathrm{~g}$ of water at 293 K ), but when heated it decomposes at 513 K
$2 \mathrm{KMnO}_{4} \rightarrow \mathrm{~K}_{2} \mathrm{MnO}_{4}+\mathrm{MnO}_{2}+\mathrm{O}_{2}$
2. Interaction between $\pi$ Bond $\&$ lone pairs on adjacent atoms
(a) Resonance
(b) Hyper conjugation
(c) Inductive Effect
(d) Electronic Effect

Answer: (a) Resonance
3. Assertion (A) : Electronegativity increase across a period

Reason (R): Effective increase in nuclear charge is more than effective shielding.

Solution: Assertion true reason true
4. S-1: Electronegativity increase down the group 14 is to pb

S-2: Group 14 contains metals, non metals and also metalloids
$\mathrm{S}-1$ is incorrect but $\mathrm{S}-2$ is correct

Column - I
a) Ziegler Natta Catalyst
b) Blood Pigment ii) CO
c) Wilkinson Catalyst
iii) Fe
d) Vitamin $B_{12}$
(a)
(b)
(c)
(d)

| $(1)$ | iv | iii | i | ii |
| :--- | :--- | :--- | :--- | :--- |
| (2) | iii | ii | i | iv |
| (3) | ii | iv | i | ii |
| $(4)$ | i | ii | iii | iv |

Key: 1
5. Appearance of Red colour on treatment with Na fusion extract of an organic compound with $\mathrm{FeSO}_{4}$ in presence of conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$ indicate element
a) N
b) Br
c) S
d) $N \& S$

Answer: (d) N \& S
6. $\mathrm{Cl}^{-}$shows disproportionation in alkaline medium:
$\mathrm{acl}_{2}+\mathrm{bOH}^{-} \rightarrow \mathrm{c} \mathrm{ClO}^{-}+\mathrm{dcl}-+\mathrm{H}^{2} \mathrm{O}$

|  | $(a)$ | $(b)$ | (c) | (d) |
| :---: | :---: | :---: | :---: | :---: |
| $(1)$ | 1 | 1 | 1 | 3 |
| $(2)$ | 3 | 6 | 2 | 4 |
| $3)$ | 1 | 2 | 1 | 1 |

4) $24 \begin{array}{llll} & 4 & 1\end{array}$

Answer: (b)
Solution: $3 \mathrm{C}_{2}+6 \mathrm{OH}^{-} \rightarrow 2 \mathrm{ClO}_{3}^{-}+4 \mathrm{Cl}-+3 \mathrm{H}_{2} \mathrm{O}$
7. The correct set of four Quantum numbers of Valence electron of $\mathrm{Rb}(37)$
a) $n=5 ., l=0 ., m=1$.,
b) $n=5 ., l=0 ., m=0 .$,
c) $n=5 ., l=1 ., m=0$.
d) $n=5 ., l=1 ., m=1$.,

Answer: b
8. Type of amino acids obtained on hydrolysis of proteins
a) $\alpha$
b) $\beta$
c) $\gamma$
d) $\delta$
9. Which of the follwing coordinatin compounds has bridging carbonyl ligand
a) $\left[\mathrm{Mn}_{2}(\mathrm{CO})_{10}\right.$
b) $\left[\mathrm{Co}_{2}(\mathrm{CO})_{8}\right]$
c) $\left[\mathrm{Cr}(\mathrm{CO})_{6}\right]$
d) $\left[\mathrm{Fe}(\mathrm{CO})_{5}\right]$
10. Calculate the Molarity of a Solution having density $=$ $1.25 \mathrm{~g} / \mathrm{ml} . \%(\mathrm{w} / \mathrm{w})$ of Solute is $31.4 \%$ of $\mathrm{H}_{2} \mathrm{SO}_{4}$ solution
a) 4
b) 9
c) 8
d) 6
11. Among the hetercyclic compound that contain sulphur atom is
a) Pyradizine
b) Furan
c) Thophene
d) Pyrrole

Key: c
12. Find the weight of Zinc formed when Zinc sulphate is electrolized for 15 min by using 0.015 Ampere current

Key : 0.0046g
13. Number of compound in which B.O = 1 and is paramagnetic $\mathrm{He}_{2}^{+}, \mathrm{O}_{2}^{+}, \mathrm{O}_{2}^{-2}, \mathrm{~N}_{2}^{+}$

Key : 0
14. Number of compounds that gives positive fehling test

Benzaldehyde, acetophenone, methanal
Key : 1
15.


Key:

16. $\mathrm{MnO}_{4}^{-}+\mathrm{KI} \xrightarrow{\text { alkine medium }} ?+\mathrm{IO}_{3}^{-}$

Key: $\mathrm{MnO}_{2}$
17.

a)


b)


c)


d)


18.

a)

b)

c)

19. Number of compunds with central atom having lone pair of electrons.

$$
\mathrm{O}_{3}, \mathrm{NH}_{3}, \mathrm{SF}_{4}, \mathrm{H}_{2} \mathrm{O}, \mathrm{XeF}_{4}
$$

20. The reactions taking place in temperature, the rate of overall reaction is given as $K=\frac{1_{2}}{k_{3}}$ then overall activate Ea for reaction, $E a_{1}=40 \mathrm{~kg} / \mathrm{mole}^{2} \mathrm{Ea}_{2}=50$ kg mole, $\mathrm{Ea}_{3}=60 \mathrm{~kg} / \mathrm{mole}$
21. Which of the follwing is correctly matches
a) Cryolite $-\mathrm{Na}_{3} \mathrm{AlF}_{6}$
b) Flurospar $\mathrm{CaF}_{2}$
c) Chloropatite $-3 \mathrm{Ca}_{3}\left(\mathrm{PO}_{4}\right)_{5} \mathrm{CaF}_{2}$
d) Carnalite $\mathrm{KCl} \cdot \mathrm{MgCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$

Key: 3

## MATHEMATICS

1. If $\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}}\left(\frac{x^{2} \cos x}{1+e^{x}}+\frac{1+\sin ^{2} x}{1+e^{\sin x^{2023}}}\right) d x=\frac{\pi}{2}(\pi+\alpha)-4$ find $\alpha$.

Ans : -12
2. A position vector of $A$ is $\overrightarrow{O A}=2 i+2 j+k, \overrightarrow{O B}=2 i+4 j+4 k$ Then angular bisector of $\angle A O B$ intersect $A B$ at point $C$. Find $\overrightarrow{O C}$.

Ans: $\left(2, \frac{8}{3}, 2\right)$
3. If $A\left[\begin{array}{ccc}1 & 0 & 0 \\ 0 & \alpha & \beta \\ 0 & \beta & \alpha\end{array}\right]$ and $|2 A|^{3}=2^{21}$. Where $\alpha, \beta \in N$

Then $(\alpha, \beta)$ can be
Ans : $(3,5)$
4. Sum of all 64 terms of G.P is 7 times of odd terms then find common ratio.

Ans: $\frac{1}{7}$
5. Let a die rolled till 2 is obtained. The probability that 2 obtained on even numbered toss is equal to :

Ans: $\frac{5}{11}$
6. Find the common ratio of a G.P whose $\mathrm{a}_{6}=2$ and $a_{1} a_{4} a_{7}$ is maximum.

Ans: $\frac{8}{5}$
7. If $a, \beta$ are the roots of $x^{2}-x+2=0$ such that $\operatorname{Im}(\alpha)>\operatorname{Im}(\beta)$, find $\alpha^{6}+\alpha^{4}+\beta^{4}-5 \alpha^{2}$

Ans : 13
8. Given $\Delta^{l e} A B C, \mathrm{y}=\mathrm{x}$ is angle bisector of $\angle A B C 2 \mathrm{x}-\mathrm{y}=2$ is equation if $\mathrm{AC}(4,6)$ and $(\alpha, \beta)$ are $\mathrm{A}, \mathrm{B}$ respectively, then $\alpha+2 \beta=$ ?
9. Evaluate : $\lim _{x \rightarrow \frac{\pi}{2}} \frac{\int_{x}^{\left(\frac{\pi}{2}\right)^{3}} \cos t^{\frac{1}{3}} d t}{\left(x-\frac{\pi}{2}\right)^{2}}$
(1) $\frac{3 \pi^{2}}{4}$
(2) $\frac{3 \pi}{4}$
(3) $\frac{3 \pi^{2}}{8}$
(4) $\frac{3 \pi}{8}$

Ans: $\frac{3 \pi^{2}}{8}$
10. $\frac{{ }^{11} c_{1}}{2}+\frac{{ }^{11} c_{2}}{3}+\ldots . . . \frac{{ }^{11} c_{9}}{10}=\frac{m}{n}$ The $m+n$ is

Ans: 2041
11. Rank of the word 'GTWENTY' in dictionary is

Ans : 553
12. $\bar{a}, \bar{b}, \bar{c}$ are three non collinear vectors $\bar{a}, 6 \bar{b}$, is collinear with $\bar{c}, \bar{b}+5 \bar{c}$ is collinear with $\bar{a}$. then $\bar{a}+\alpha \vec{b}+\vec{\beta} \bar{c}=0$ then find $\alpha+\beta=$ ?

Ans: 36
13. If $f(x)=\int \frac{\operatorname{cosec} x+\sin x}{\operatorname{cosec} x \sec x+\tan x \sin ^{2} x} d x$
and $\lim _{x \rightarrow \frac{\pi}{2}} f(x)=1$, then find $f\left(\frac{\pi}{4}\right)=$ ?
14. Let $A$ be square matrix such that $A A^{\top}=I$. Then $\left.\frac{1}{2} A\left[\left(A+A^{\prime}\right)\right]^{2}+\left(A-A^{\prime}\right)^{2}\right]$ equal to
(1) $A^{3}+A^{\top}$
(2) $A^{2}+A^{\top}$
(3) $A^{2}+1$
(4) $\mathrm{A}^{3}+\mathrm{I}$.

Ans: $A^{3}+A^{\top}$
15. If $z=\frac{1}{2}-2 i$ such that
$|z+1|=\alpha z+\beta(1+i), i=\sqrt{-1}$ and $\quad \alpha, \beta \in N$ then $\alpha+\beta$ is

Ans :
16. Let R be a relation on $z \times z$ defined by $(a, b)(c, d)$ if adbc is divisible by 5 is what type of relation?
17. Domain of $f(x)=\frac{\log (x-1)}{\log _{x-1}(x-4)}$ is

Ans: $(4, \infty)$
18. If $\frac{d y}{d x}-\left(\frac{\sin 2 x}{1+\cos ^{2} x}\right) y=\frac{\sin x}{1+\cos ^{2} x}$ and $y(0)=0$ then
$y\left(\frac{\pi}{2}\right)$ is
Ans: 1

Memory based Questions

Moderate

## Memory based Questions

| Chapter | Grade | No. of Questions | Difficulty Level |
| :--- | :--- | :--- | :--- |
| Units \& Measurement | 11 | 1 | Easy |
| Motion in a straight line | 11 | 1 | Moderate |
| Motion in a plane | 11 | 1 | Easy |
| Laws of motion | 11 | 2 | Moderate, Easy |
| Work, energy and power | 11 | 2 | Moderate, Easy |
| System of particles and <br> rotational motion | 11 | 1 | Moderate |
| Gravitation | 11 | 1 | Moderate |
| Mechanical properties of <br> solids | 11 |  | Moderate |
| Mechanical properties of <br> fluids | 11 | 1 |  |
| Thermal properties of matter | 11 |  |  |

## Memory based Questions

| Chapter | Grade | No. of Questions | Difficulty Level |
| :--- | :--- | :--- | :--- |
| Thermodynamics | 11 | 1 | Easy |
| Kinetic theory of gases | 11 | 1 | Easy |
| Oscillations | 11 | 1 | Easy |
| Waves | 11 |  | Moderate |
| Electrostatic potential and <br> capacitance | 12 | 1 | Difficult |
| Electric charges and fields | 12 | 1 | Easy, Difficult |
| Current Electricity | 12 | 2 | Moderate |
| Moving charges and <br> magnetism | 12 | 1 | Moderate |
| Magnetism and matter | 12 |  |  |
| Electromagnetic induction | 12 | 1 |  |

## Memory based Questions

| Chapter | Grade | No. of Questions | Difficulty Level |
| :--- | :--- | :--- | :--- |
| Alternating Current | 12 | 1 | Difficult |
| Electromagnetic waves | 12 |  |  |
| Ray optics and optical instruments | 12 | 2 | Moderate |
| Wave optics | 12 |  |  |
| Dual nature of radiation and matter | 12 | 1 | Moderate |
| Atoms | 12 | 1 | Moderate |
| Nuclei | 12 |  |  |
| Semiconductor electronics: Materials; <br> devices and simple circuits | 12 |  |  |

Memory based Questions


## Memory based Questions

| Chapter | Grade | No. of Questions | Difficulty Level |
| :--- | :--- | :--- | :--- |
| Sets and Relations | 11 | 1 | Easy |
| Functions | 11 | 1 | Easy |
| Trigonometric Functions | 11 |  |  |
| Principle of Mathematical <br> Induction, Linear inequalities | 11 |  |  |
| Quadratic Equations | 11 |  | Easy |
| Complex Numbers | 11 | 2 | Easy |
| Permutations and <br> Combinations | 11 | 1 | easy |
| Binomial Theorem | 11 | 1 | Moderate |
| Sequences and Series | 11 | 1 | Moderate |
| Straight Lines | 11 | 1 |  |

Memory based Questions

| Chapter | Grade | No. of Questions | Difficulty Level |
| :--- | :--- | :--- | :--- |
| Circles | 11 |  |  |
| Conic Section | 11 |  |  |
| Limits and Derivatives | 11 |  | Easy |
| Statistics | 12 |  |  |
| Matrices | 12 | 2 |  |
| Determinants | 12 |  | Difficult |
| Inverse Trigonometric <br> Functions | 12 |  | Moderate |
| Continuity and <br> Differentiability | 12 |  |  |
| Applications of Derivatives | 12 | 1 | 1 |

Memory based Questions

| Chapter | Grade | No. of Questions | Difficulty Level |
| :--- | :--- | :--- | :--- |
| Definite Integration | 12 | 2 | Moderate |
| Applications of Integrals | 12 |  |  |
| Differential Equations | 12 |  |  |
| Vectors Algebra | 12 | 2 | Easy |
| Three Dimensional Geometry | 12 |  |  |
| Probability | 12 | 1 | Moderate |
| Trigonometric Equations | 11 |  |  |

## Memory based Questions

## Difficult

15\%

## 29th Jan. Shift - 1 Chemistry Analysis

Memory based Questions

| Chapter | No. of Questions | Difficulty Level |
| :--- | :--- | :--- |
| Some basic concepts of chemistry | 1 | Moderate |
| Structure of Atom | 1 | Moderate |
| Periodic Classification |  |  |
| Chemical bonding and molecular <br> structure | 3 | Moderate, Easy |
| States of matter |  |  |
| Thermodynamics | 3 | Moderate, Easy |
| Equilibrium | 2 | Easy |
| Redox reactions | 2 | Difficult, Moderate |
| p - block elements |  |  |
| Hydrocarbons |  |  |

## 29th Jan. Shift - 1 Chemistry Analysis

Memory based Questions

| Chapter | No. of Questions | Difficulty Level |
| :--- | :--- | :--- |
| Solution |  |  |
| Electrochemistry | 1 | Moderate |
| Chemical kinetics | 1 | Moderate |
| General principles and processes of <br> isolation of elements |  |  |
| d \& f block elements | 1 | Moderate |
| Coordination compounds | 1 | Moderate |
| Haloalkanes and Haloarenes | 1 | Difficult |
| Alcohol, Phenol Ether | 2 | Moderate |
| Aldehyde, Ketone, Carboxylic acid |  |  |
| Amines |  |  |

## 29th Jan. Shift - 1 Chemistry Analysis

## Memory based Questions

| Chapter | No. of Questions | Difficulty Level |
| :--- | :--- | :--- |
| Biomolecule | 1 | Easy |
| Organic Chemistry-Some basic <br> principles and techniques(General <br> Organic) | 2 | Easy |
| Salt Analysis |  |  |

## HISTORY CREATED

## SRI CHAITANYA STUDENTS SECURE TOP RANKS in JEE ADVANCED 2023

## ALL-INDIA OPEN CATEGORY RANKS



